

## Regioselective synthesis of diaryl sulfides by [3+3] cyclizations of 1,3-bis(trimethylsilyloxy)-1,3-dienes

Muhammad A. Rashid,<sup>a</sup> Helmut Reinke<sup>a</sup> and Peter Langer<sup>a,b,\*</sup>

<sup>a</sup>*Institut für Chemie, Universität Rostock, Albert-Einstein-Str. 3a, 18059 Rostock, Germany*

<sup>b</sup>*Leibniz-Institut für Katalyse e. V. an der Universität Rostock, Albert-Einstein-Str. 29a, 18059 Rostock, Germany*

Received 21 December 2006; revised 23 January 2007; accepted 26 January 2007

Available online 1 February 2007

**Abstract**—Functionalized and sterically encumbered diaryl sulfides were prepared based on [3+3] cyclizations of 1,3-bis(trimethylsilyloxy)-1,3-dienes.

© 2007 Elsevier Ltd. All rights reserved.

Diaryl sulfides are of considerable pharmacological relevance and occur in a number of natural products. These include various dibenzothiophenes,<sup>1</sup> highly cytotoxic lissoclinotoxins (varacins),<sup>2</sup> lissoclibadins (isolated from *Lissoclinum cf. badium*),<sup>3</sup> cyclo(penta-1,4-phenylene sulfide), cyclotetra(*p*-phenylene sulfide),<sup>4</sup> and a number of natural products isolated from *Streptomyces griseus*.<sup>5</sup> Diaryl sulfides have been prepared by thermal reaction of arenes with sulfur.<sup>6</sup> However, these reactions often suffer from low regioselectivity, harsh reaction conditions and formation of polysulfides. Diaryl sulfides are available also by condensation of Grignard or organolithium compounds with chlorophenyl sulfide<sup>7</sup> and by base-mediated reaction of chloroarenes with thiophenols.<sup>8</sup> Some years ago, Chan and co-workers reported<sup>9</sup> the synthesis of salicylates by formal [3+3] cyclization of 1,3-bis(trimethylsilyloxy)-1,3-dienes<sup>10</sup> with 3-siloxy-2-en-1-ones. In recent years, we have reported the application of this method to the synthesis of a variety of functionalized arenes.<sup>11</sup> Chan and Prasad reported the synthesis of 2-(thiophenoxy)benzoates based on [3+3] cyclizations of 1-trimethylsilyloxy-1-methoxy-3-thiophenoxy-1,3-butadiene.<sup>12</sup> Herein, we report what is, to the best of our knowledge, the first synthesis of diaryl sulfides by [3+3] cyclizations of 1,3-bis(silyl enol ethers). These reactions provide a convenient and regioselective approach to sterically

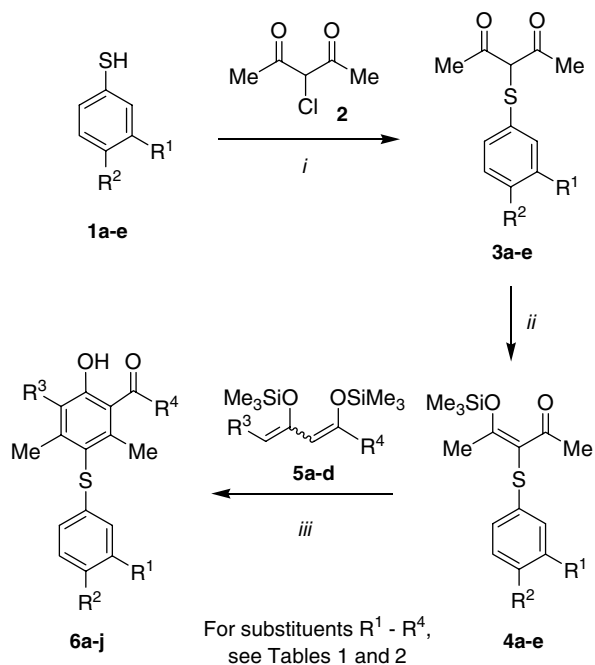
encumbered and functionalized diaryl sulfides which are not readily available by other methods. In contrast to the coupling reactions outlined above, our method relies on the assembly of one of the two arene moieties.

The reaction of 3-chloropentane-2,4-dione (**2**) with thiophenols **1a–e** afforded, following a known procedure,<sup>13</sup> 3-(thiophenoxy)pentane-2,4-diones **3a–e**. Silylation of the latter gave the novel 2-thiophenoxy-3-silyloxy-2-en-1-ones **4a–e** (Scheme 1, Table 1). The TiCl<sub>4</sub> mediated [3+3] cyclization of **4a–e** with 1,3-bis(trimethylsilyloxy)-1,3-dienes **5a–d**—readily available from methyl acetoacetate, ethyl 3-oxopentanoate, ethyl 3-oxohexanoate, and ethyl acetoacetate<sup>9</sup>—afforded diaryl sulfides **6a–j** (Scheme 1, Table 1).<sup>14</sup> During the optimization of the cyclization reaction, the (high) concentration, the temperature, and the choice of Lewis acid proved to be important parameters. In fact, we have earlier found for related reactions that the use of SnCl<sub>4</sub> or BF<sub>3</sub>·OEt<sub>2</sub> resulted in a decrease of yield. No conversion was observed when Me<sub>3</sub>SiOTf was employed. All structures were established by spectroscopic methods. The structure of **6a** was independently confirmed by X-ray crystal structure analysis (Fig. 1).<sup>15</sup>

The triethylamine mediated reaction of ethyl 4-chloroacetoacetate with thiophenol (**1a**) afforded ethyl 4-(thiophenoxy)acetoacetate (**7a**),<sup>16</sup> which was transformed in two steps into novel thiophenoxy-substituted 1,3-bis(trimethylsilyloxy)-1,3-diene **9a** (Scheme 2, Table 2). Likewise, **9b** was prepared from ethyl 4-chloroacetoacetate and 4-methoxythiophenol (**1b**). The TiCl<sub>4</sub> mediated

**Keywords:** Arenes; Cyclizations; Diaryl sulfides; Silyl enol ethers.

\* Corresponding author. Tel.: +49 381 4986410; fax: +49 381 4986412; e-mail: [peter.langer@uni-rostock.de](mailto:peter.langer@uni-rostock.de)

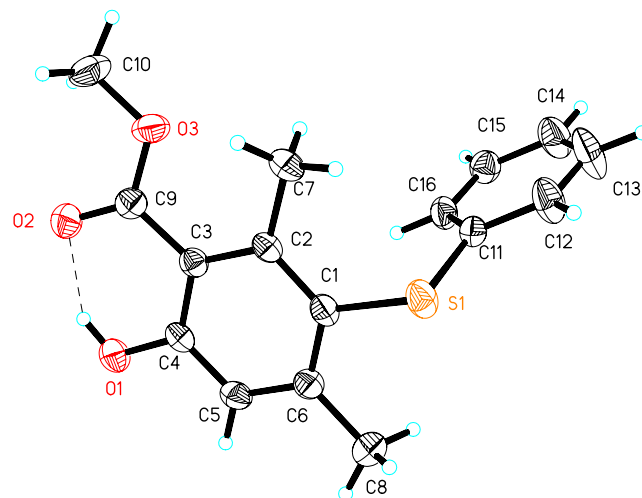


**Scheme 1.** Synthesis of **6a–j**: Reagents and conditions: (i): C<sub>3</sub>H<sub>5</sub>N/C<sub>5</sub>H<sub>11</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, MeOH, 0–20 °C, 6 h; (ii): Me<sub>3</sub>SiCl, NEt<sub>3</sub>, C<sub>6</sub>H<sub>6</sub>, 20 °C, 72 h; (iii): TiCl<sub>4</sub>, CH<sub>2</sub>Cl<sub>2</sub>, –78–20 °C, 20 h.

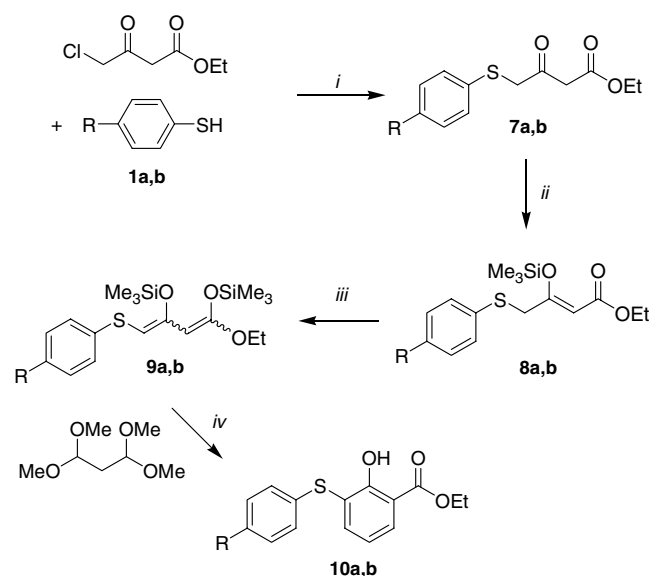
[3+3] cyclization of **9a** and **9b** with 1,1,3,3-tetramethoxypropane afforded diaryl sulfides **10a** and **10b**, respectively.

The [3+3] cyclization of 1,3-bis(trimethylsilyloxy)-1,3-dienes **9a** and **9b** with 3-silyloxy-2-en-1-ones **11a** and **11b**—prepared from pentane-2,4-dione and heptane-3,4-dione—afforded diaryl sulfides **12a** and **12b**, respectively (Scheme 3, Table 3). Bis(diaryl sulfides) **12c** and **12d** were prepared by cyclization of **9a** and **9b** with 2-thiophenoxy-3-silyloxy-2-en-1-one **4a**, respectively.

In conclusion, we have reported the synthesis of functionalized and sterically encumbered diaryl sulfides by [3+3] cyclizations of 4-thiophenoxy-1,3-bis(trimethylsilyloxy)-1,3-dienes with 3-silyloxy-2-en-1-ones and 1,1,3,3-tetramethoxypropane and by [3+3] cyclization of 1,3-bis(trimethylsilyloxy)-1,3-dienes with 2-thiophenoxy-3-silyloxy-2-en-1-ones.



**Figure 1.** ORTEP plot of **6a**.



**Scheme 2.** Synthesis of **10a,b**: Reagents and conditions: (i): NEt<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 30 min, 0 °C; (ii): Me<sub>3</sub>SiCl, NEt<sub>3</sub>, C<sub>6</sub>H<sub>6</sub>, 20 °C, 72 h; (iii): (1) LDA, THF, –78 °C, 1 h, (2) Me<sub>3</sub>SiCl, –78–20 °C; (iv): TiCl<sub>4</sub>, CH<sub>2</sub>Cl<sub>2</sub>, –78–20 °C, 20 h.

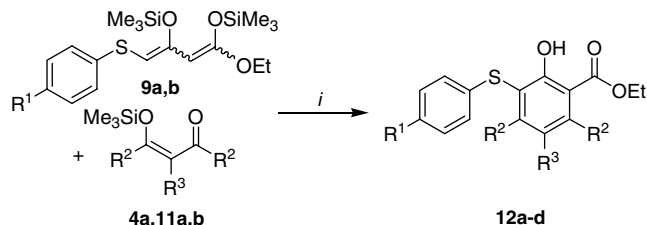
**Table 1.** Synthesis of **6a–j**

3,4	5	6	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	R <sup>4</sup>	% (3) <sup>a</sup>	% (4) <sup>a</sup>	% (6) <sup>a</sup>
a	a	a	H	H	H	OMe	72	81	48
a	b	b	H	H	Me	OEt			40
a	c	c	H	H	Et	OEt			40
b	d	d	H	OMe	H	OEt	33	90	43
b	b	e	H	OMe	Me	OEt			35
b	c	f	H	OMe	Et	OEt			38
c	a	g	H	Br	H	OMe	28	79	36
d	b	h	H	Me	Me	OEt	81	92	33
e	a	i	OMe	H	H	OMe	73	81	32
e	b	j	OMe	H	Me	OEt			30

<sup>a</sup> Yields of isolated products.

**Table 2.** Synthesis of **10a,b**

7–10	R	% (7) <sup>a</sup>	% (8) <sup>a</sup>	% (9) <sup>a</sup>	% (10) <sup>a</sup>
<b>a</b>	H	80	84	89	31
<b>b</b>	OMe	81	78	85	30

<sup>a</sup> Yields of isolated products.

**Scheme 3.** Synthesis of **12a–d**: Reagents and conditions: (i): TiCl<sub>4</sub>, CH<sub>2</sub>Cl<sub>2</sub>, -78→20 °C, 20 h.

**Table 3.** Synthesis of **12a–d**

4a,11	9	12	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	% (12) <sup>a</sup>
<b>11a</b>	<b>a</b>	<b>a</b>	H	Me	H	48
<b>11b</b>	<b>b</b>	<b>b</b>	OMe	Et	H	35
<b>4a</b>	<b>a</b>	<b>c</b>	H	Me	PhS	34
<b>4a</b>	<b>b</b>	<b>d</b>	OMe	Me	PhS	34

<sup>a</sup> Yields of isolated products.

### Acknowledgment

Financial support by the state of Pakistan (HEC scholarship for M.A.R.) is gratefully acknowledged.

### References and notes

- See for example: Mori, Y.; Taneda, S.; Hayashi, H.; Sakushima, A.; Kamata, K.; Suzuki, A. K.; Yoshino, S.; Sakata, M.; Sagai, M.; Seki, K.-i. *Biol. Pharm. Bull.* **2002**, *25*, 145.
- (a) Davidson, B. S.; Molinski, T. F.; Barrows, L. R.; Ireland, C. M. *J. Am. Chem. Soc.* **1991**, *113*, 4709; (b) Behar, V.; Danishefsky, S. J. *J. Am. Chem. Soc.* **1993**, *115*, 7017; (c) Toste, F. D.; Still, I. W. J. *J. Am. Chem. Soc.* **1995**, *117*, 7261.
- (a) Davis, R. A.; Sandoval, I. T.; Concepcion, G. P.; Moreira da Rocha, R.; Ireland, C. M. *Tetrahedron* **2003**, *59*, 2855; (b) Liu, H.; Fujiwara, T.; Nishikawa, T.; Mishima, Y.; Nagai, H.; Shida, T.; Tachibana, K.; Kobayashi, H.; Mangindaan, R. E. P.; Namikoshi, M. *Tetrahedron* **2005**, *61*, 8611.
- Kaplan, M. L.; Reents, W. D. *Tetrahedron Lett.* **1982**, *23*, 373.
- Hosoya, Y.; Adachi, H.; Nakamura, H.; Nishimura, Y.; Naganawa, H. *Tetrahedron Lett.* **1996**, *37*, 9227.
- See for example: (a) Dougherty, G.; Hammond, P. D. *J. Am. Chem. Soc.* **1935**, *57*, 117; (b) Glass, H. B.; Reid, E. E. *J. Am. Chem. Soc.* **1929**, *51*, 3428; for the trifluoromethanesulfonic acid catalyzed sulfurization of cycloalkanes, see: (c) Olah, G. A.; Wang, Q.; Prakash, G. K. S. *J. Am. Chem. Soc.* **1990**, *112*, 3697.
- See for example: Chua, M.; Hoyer, H. Z. *Naturforsch. B* **1965**, *20*, 416.
- (a) Campbell, J. R. *J. Org. Chem.* **1964**, *29*, 1830; (b) Baxter, I.; Ben-Haida, A.; Colquhoun, H. M.; Hodge, P.; Kohnke, F. H.; Williams, D. J. *Chem. Eur. J.* **2000**, *6*, 4285.
- (a) Chan, T.-H.; Brownbridge, P. *J. Am. Chem. Soc.* **1980**, *102*, 3534; (b) Brownbridge, P.; Chan, T.-H.; Brook, M. A.; Kang, G. J. *Can. J. Chem.* **1983**, *61*, 688.
- For a review of 1,3-bis(silyl enol ethers), see: Langer, P. *Synthesis* **2002**, 441.
- (a) Dede, R.; Langer, P. *Tetrahedron Lett.* **2004**, *45*, 9177; (b) Nguyen, V. T.; Appel, B.; Langer, P. *Tetrahedron* **2006**, *62*, 7674; (c) Ahmed, Z.; Fischer, C.; Spannenberg, A.; Langer, P. *Tetrahedron* **2006**, *62*, 4800; (d) Nguyen, V. T. H.; Bellur, E.; Appel, B.; Langer, P. *Synthesis* **2006**, 1103; (e) Mroß, G.; Langer, P. *Tetrahedron Lett.* **2006**, *47*, 8519; (f) Ahmed, Z.; Langer, P. *Synlett* **2006**, 3361; (g) Mamat, C.; Pundt, T.; Schmidt, A.; Langer, P. *Tetrahedron Lett.* **2006**, *47*, 2183.
- Chan, T. H.; Prasad, C. V. C. *J. Org. Chem.* **1986**, *51*, 3012.
- Yoshida, Z.; Ogoshi, H.; Tokumitsu, T. *Tetrahedron* **1970**, *26*, 2987.
- General procedure for the synthesis of diaryl sulfides 6a–j, 10a,b, and 12a–d*: To a dichloromethane solution (2 mL/mmole) of **4**, **11**, or 1,1,3,3-tetramethoxypropane (1.0 mmole) and of **5** or **9** (1.0 mmole) was added TiCl<sub>4</sub> (1.0 mmole) at -78 °C. The solution was allowed to warm to ambient temperature within 20 h. To the solution was added a saturated solution of NaHCO<sub>3</sub> (15 mL). The organic and the aqueous layer were separated and the latter was extracted with diethyl ether (3 × 20 mL). The filtrate was concentrated in vacuo and the residue was purified by chromatography (silica gel, EtOAc/n-heptane = 1:4). *Synthesis of methyl 4,6-dimethyl-5-(thiophenoxy) salicylate (6a)*: Starting with 3-(silyloxy)alk-2-en-1-one **4a** (200 mg, 0.71 mmole), 1,3-bis(silyl enol ether) **5a** (185 mg, 0.71 mmole), and TiCl<sub>4</sub> (0.08 mL, 0.71 mmole), **6a** was isolated as a colorless solid (99 mg, 48%), mp 77 °C; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>): δ 2.40 (s, 3H, CH<sub>3</sub>), 2.72 (s, 3H, CH<sub>3</sub>), 3.95 (s, 3H, OCH<sub>3</sub>), 6.89 (d, 2H, J = 8.2 Hz, ArH), 6.76 (s, 1H, ArH), 6.91 (s, 1H, ArH), 7.05 (br t, 1H, J = 7.2 Hz, ArH), 7.18 (br t, 1H, J = 7.4 Hz, ArH), 11.17 (s, 1H, OH); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): δ 21.4, 2.7, 52.3, (CH<sub>3</sub>), 112.4 (C), 117.7 (CH), 122.7 (C), 124.6 (CH), 125.2 (2C, CH), 128.9 (2C, CH), 138.2, 147.1, 151.4, 162.5, 171.8 (C); IR (KBr): ν̄ = 3061 (m), 2954 (m), 1663 (s), 1478 (s), 1360 (s), 1233 (s), 1187 (m), 1024 (m), 947 (w), 740 (s), 690 (m), 629 (w) cm<sup>-1</sup>; GC-MS (EI, 70 eV): m/z (%): 288.1 (M<sup>+</sup>, 57), 256.1 (100), 185.1 (7), 91 (6). Elemental Anal. Calcd (%) for C<sub>16</sub>H<sub>16</sub>O<sub>3</sub>S (288.08): C, 66.64; H, 5.59. Found: C, 66.81; H, 5.68.
- CCDC 632304 contains all crystallographic details of this publication which are available free of charge at [www.ccdc.cam.ac.uk/conts/retrieving.html](http://www.ccdc.cam.ac.uk/conts/retrieving.html) or can be ordered from the following address: Cambridge Crystallographic Data Centre, 12 Union Road, GB-Cambridge CB21EZ; Fax: (+44)1223 336 033; or deposit@ccdc.cam.ac.uk.
- (a) Sircar, I.; Gregor, E. K.; Anderson, K. R.; Haleen, S. J.; Shih, Y.-H.; Weishaar, R. E.; Steffen, R. P.; Pugsley, T. A.; Taylor, M. D. *J. Med. Chem.* **1991**, *34*, 2248; (b) Shimada, K.; Kaburagi, Y.; Fukuyama, T. *J. Am. Chem. Soc.* **2003**, *125*, 4048.